

ZnO/Au Hybrid Layered Structure on Silk Textile Prepared by Supercritical CO₂ Promoted Electrodeposition for Applications in Flexible Multifunction Electronic Devices

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Motivation & Introduction

1. Functional wearable devices

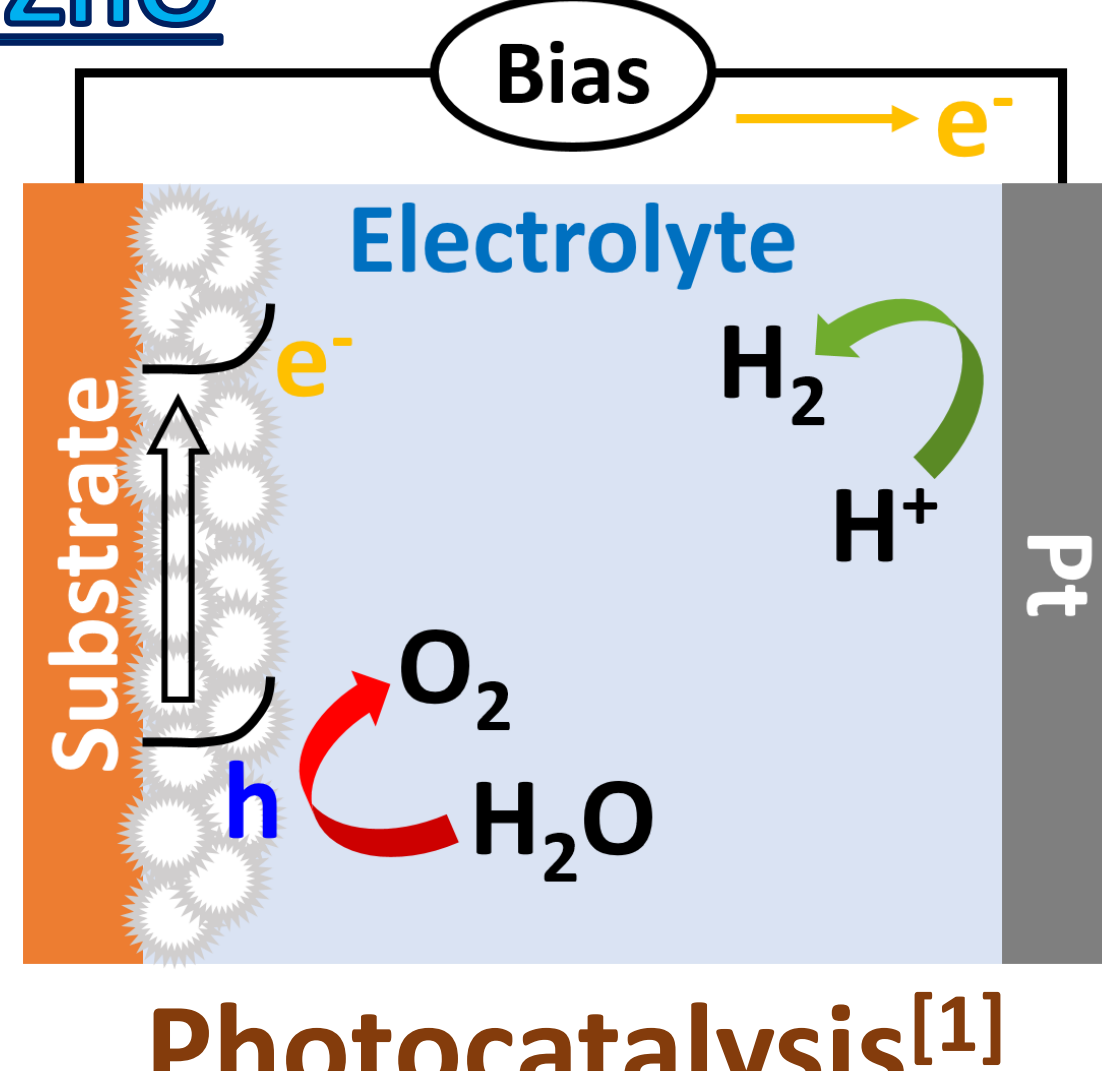


2. Merits of Au [2]

- Biocompatibility
 - Ductility
 - Corrosion
- Stretchability
- Biocompatibility

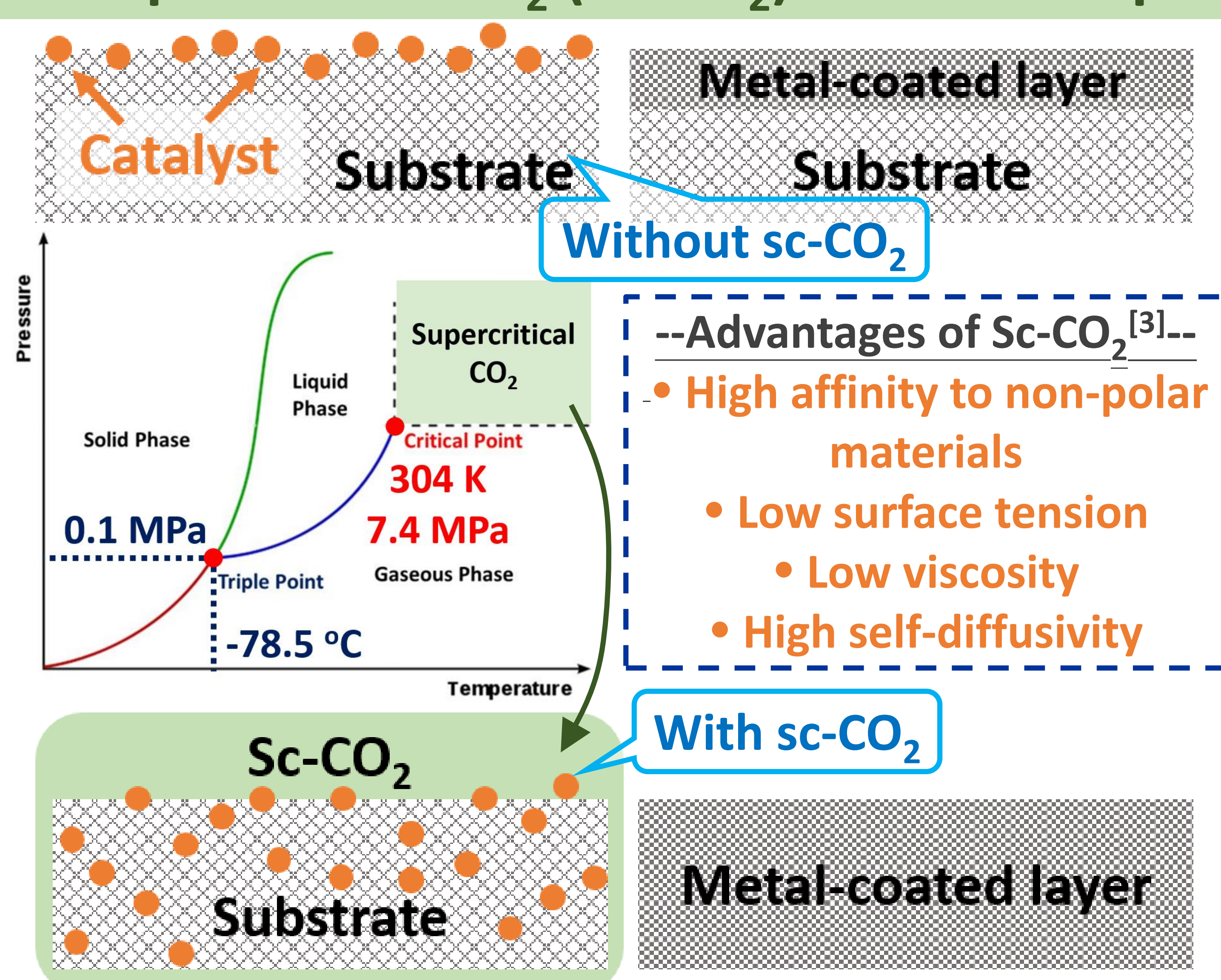
3. Silk

4. ZnO



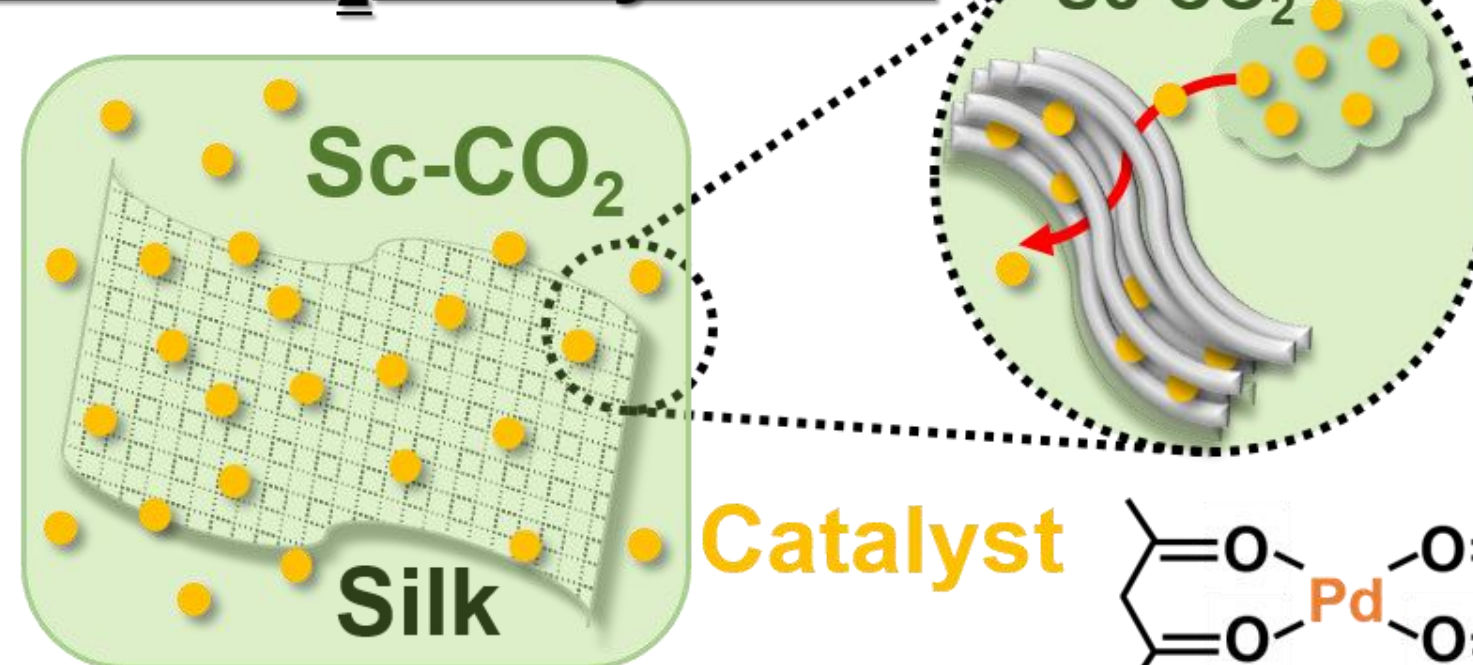
Photocatalysis^[1]

Supercritical CO₂ (Sc-CO₂) assisted step

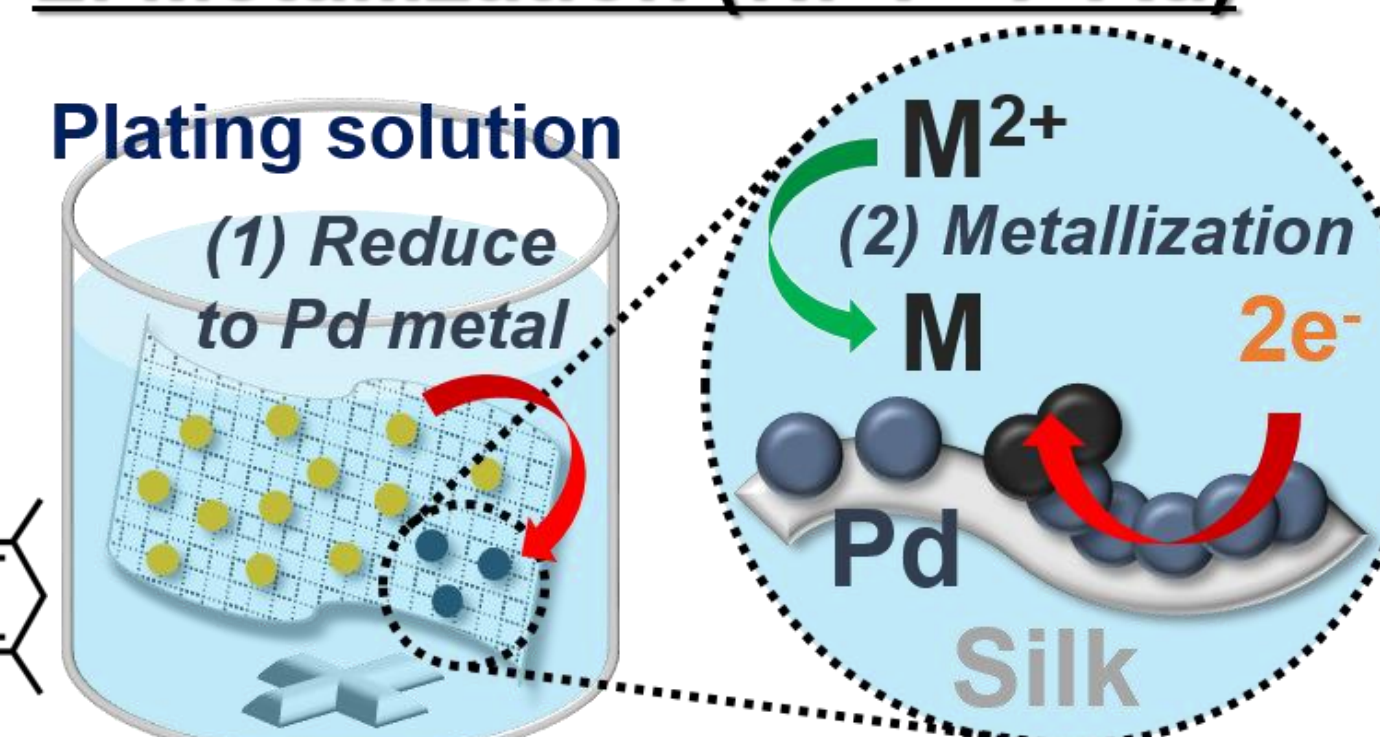


Experimental procedures

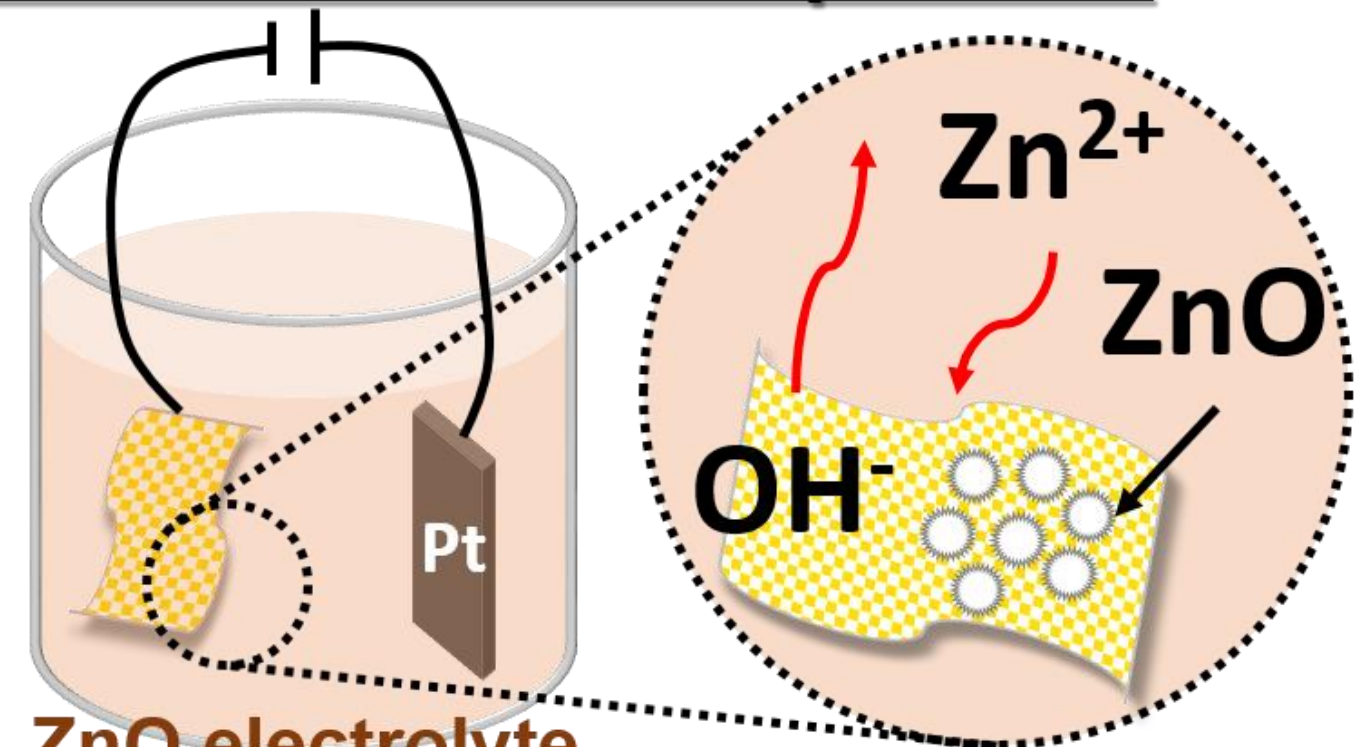
1. Sc-CO₂ catalyzation



2. Metallization (Ni-P → Au)



3. Cathodic ZnO deposition



4. Evaluations

- Surface morphology
- Crystal structure
- Electrical conductivity
- Adhesive firmness
- Photocatalytic activity

Results and Discussion I – Metallization

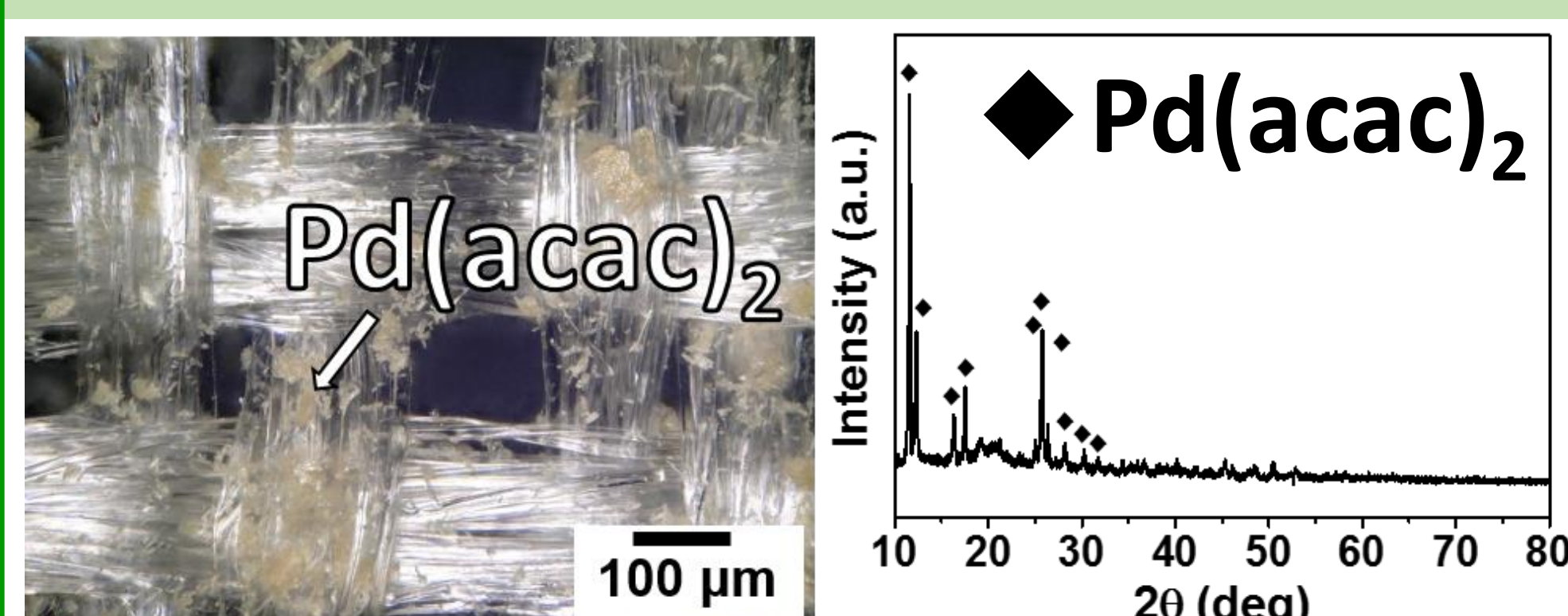


Fig. 1 (a) OM image and (b) XRD pattern of silk textile after sc-CO₂ catalyzation.

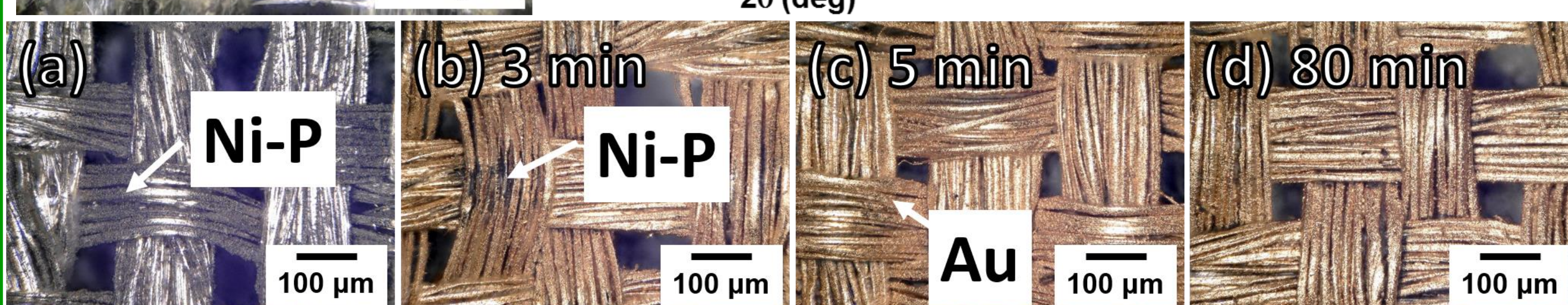


Fig. 2 Au metallization at (a) 0, (b) 3, (c) 5, and (d) 80 min.

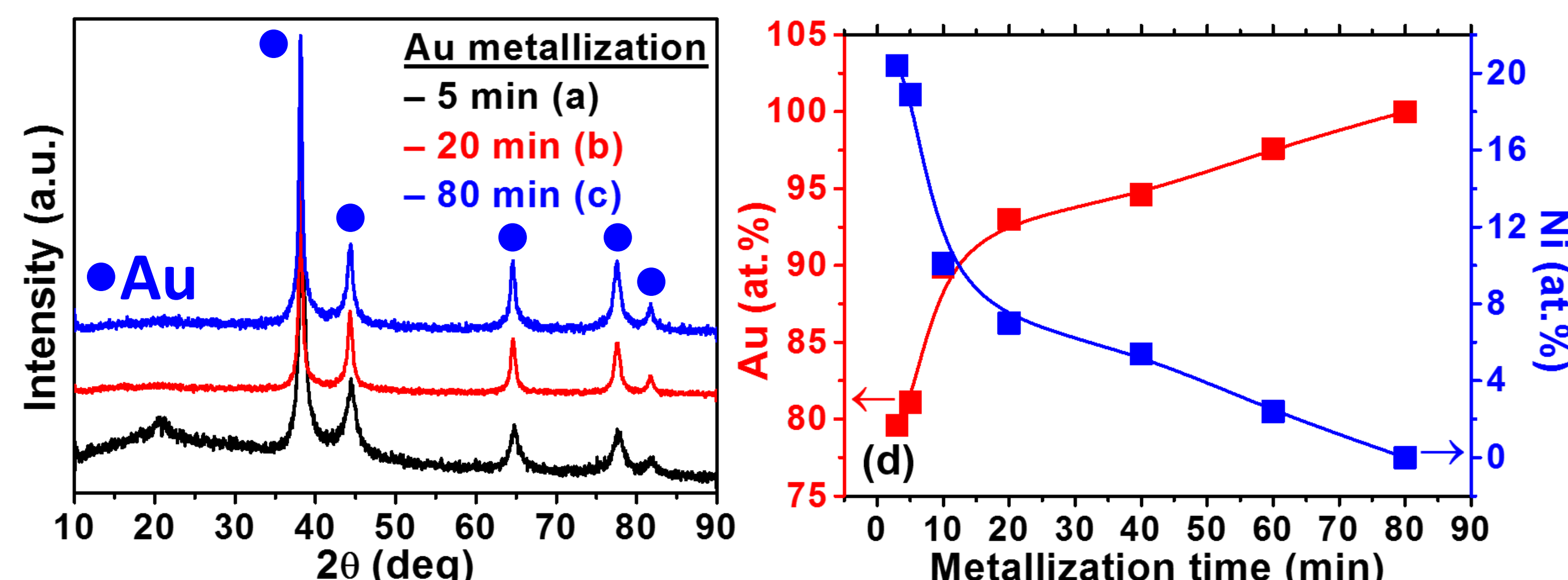


Fig. 3 XRD patterns of Au/silk at the metallization time of (a) 5, (b) 20, and (c) 80 min and (d) composition of Au/silk composite.

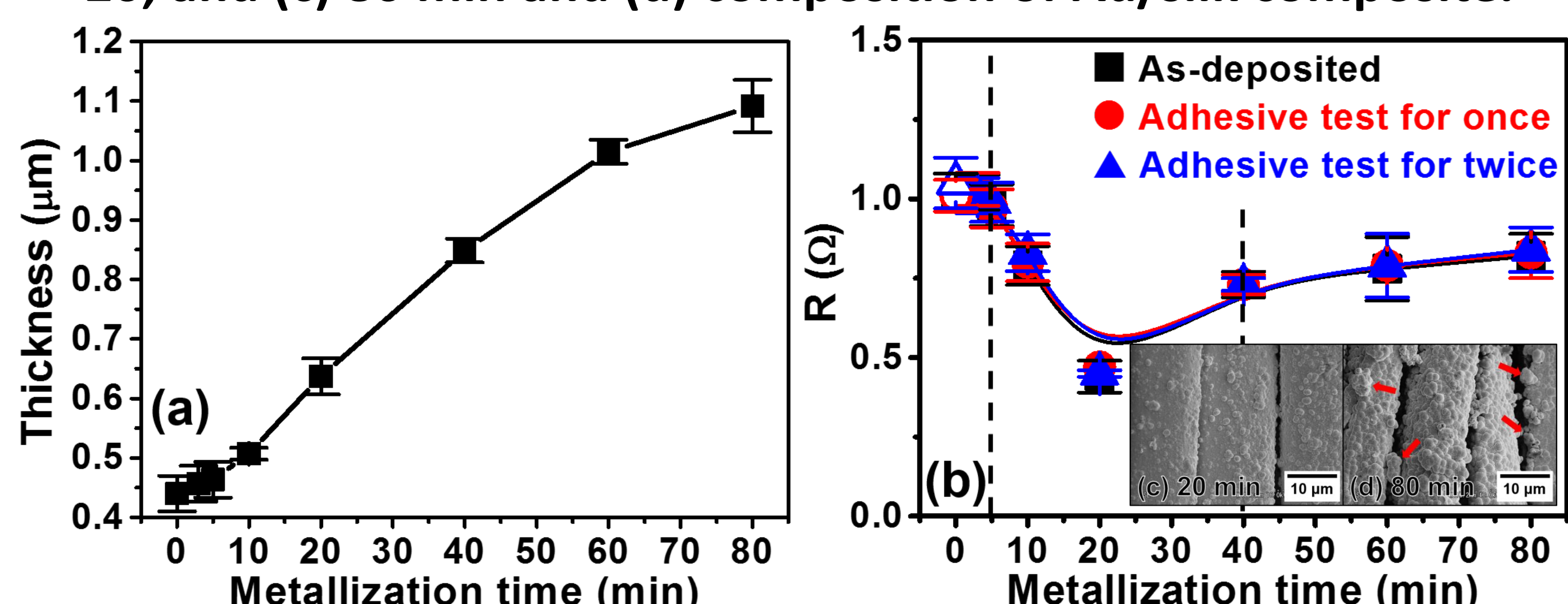


Fig. 4 (a) Au thickness – metallization time, (b) electrical resistance of Au/silk, and SEM images at metallization of (c) 20 and (d) 80 min.

Results and Discussion II – ZnO deposition

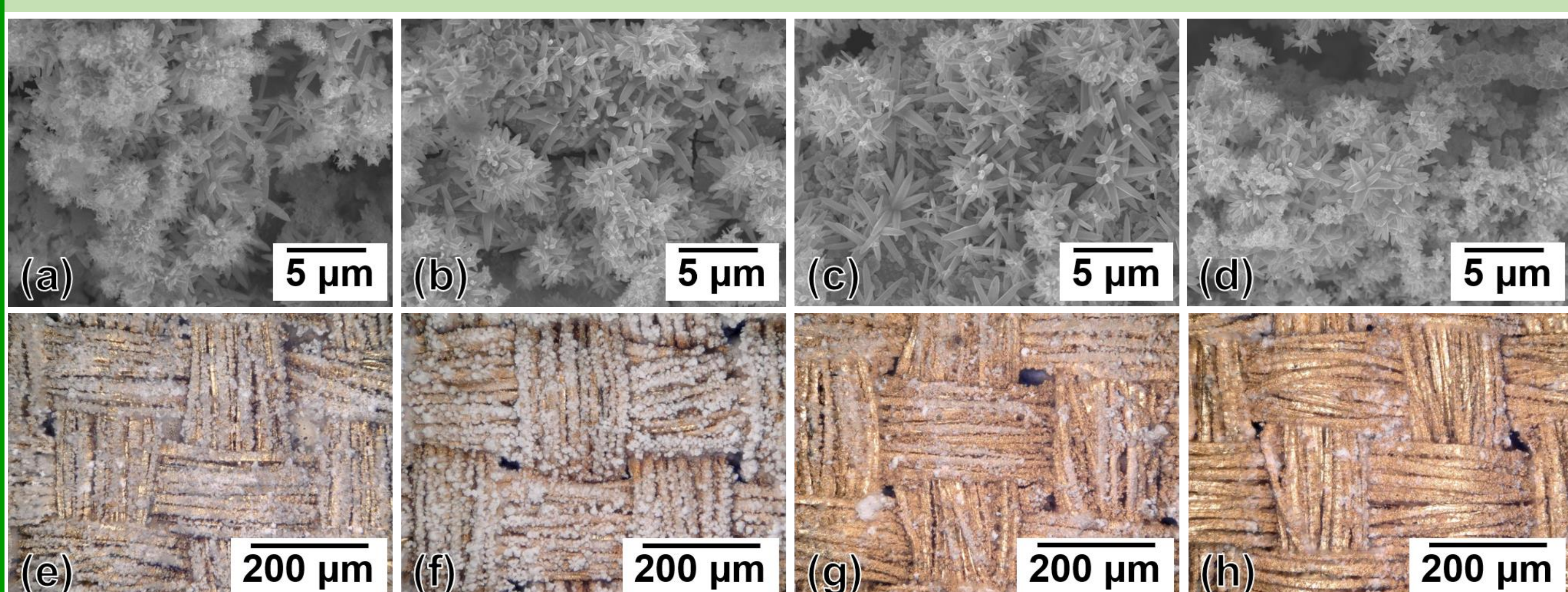


Fig. 5 ZnO deposited SEM images on Au/silk with (a) 0 wt.%, (b) 0.1 wt.%, (c) 0.2 wt.%, and (d) 0.3 wt.% and OM images on Au/silk with (a) 0 wt.%, (b) 0.1 wt.%, (c) 0.2 wt.%, and (d) 0.3 wt.% [H₂O₂].

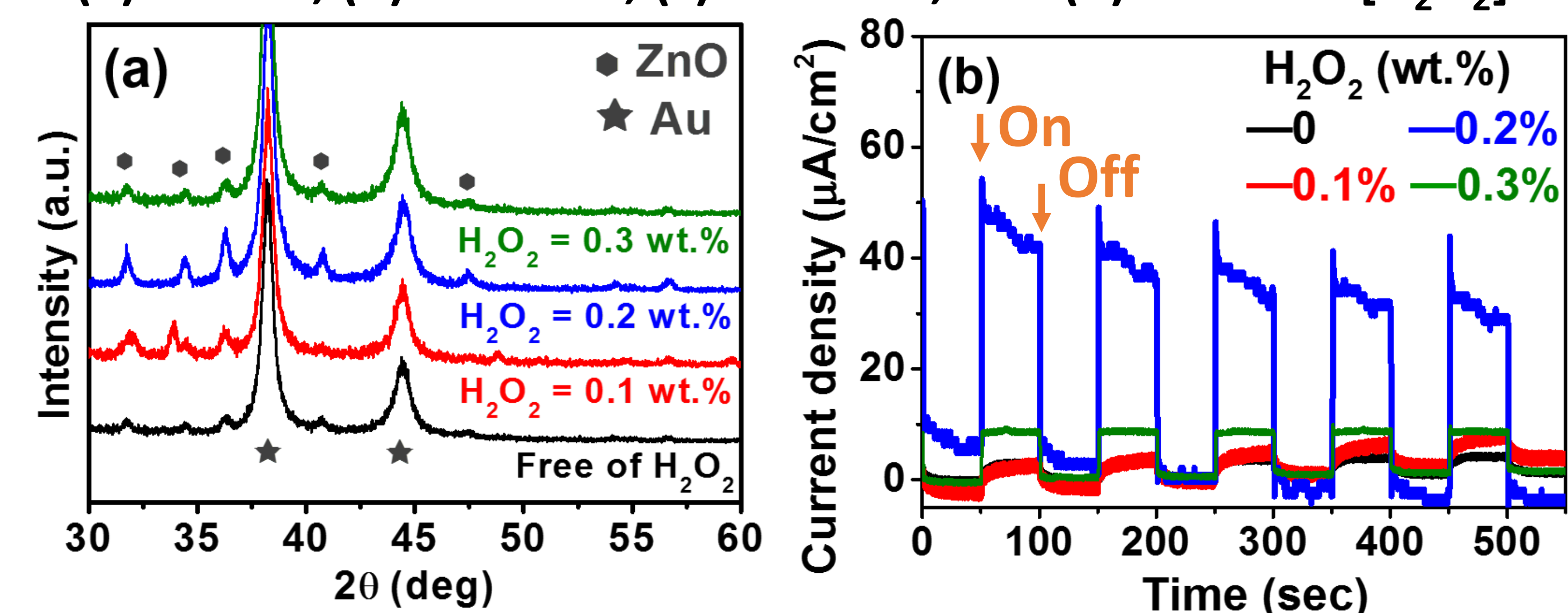


Fig. 6 (a) XRD patterns and (b) photocurrent density of the ZnO/Au/silk hybrid materials in 0.5 M Na₂SO₄ with various [H₂O₂].

Conclusions

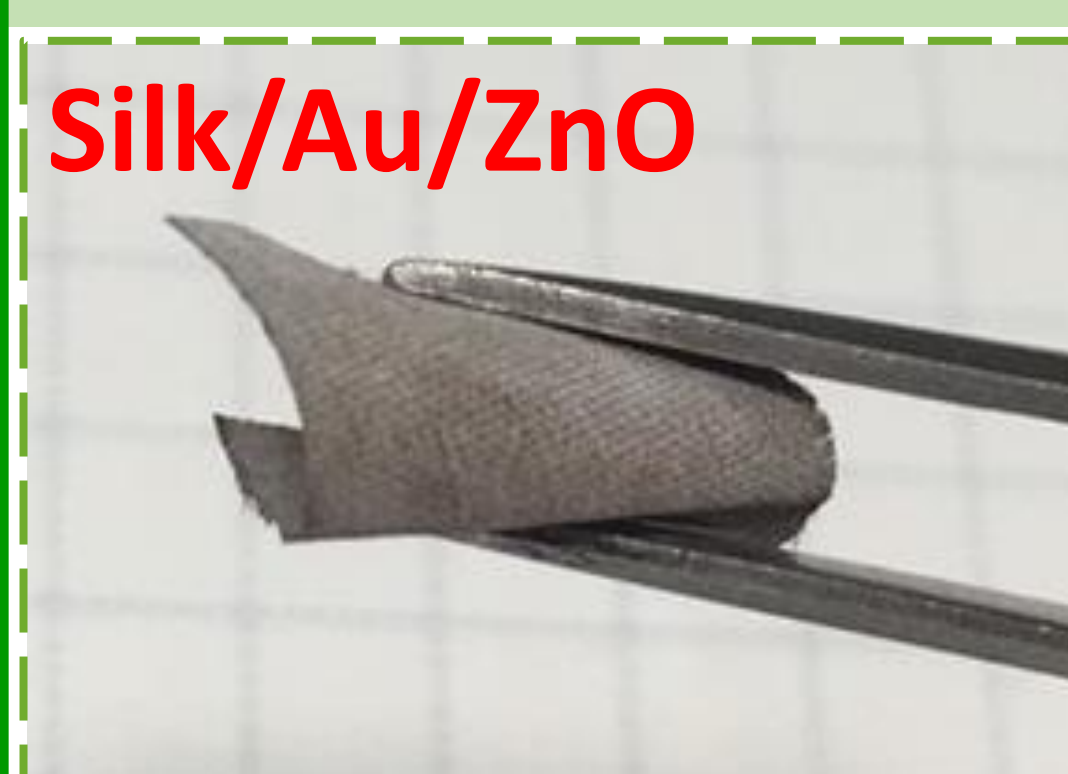


Fig. 7 Silk/Au/ZnO composite material

1. Flexible, conductive, and photocatalytic composite Silk/Au/ZnO materials.
2. Integration of ZnO (functional material) and silk textile (cloth material) together to make the materials close to daily life.
3. Photocurrent density enhanced 11.5 times with the optimized deposition electrolyte by cathodic deposition.

Ref: [1] B.R. Sankapal, H.B. Gajare, S.S. Karade, R.R. Salunkhe, D.P. Dubal, Electrochim. Acta 192 (2016) 377-384.

[2] H.M. Lee, M. Ge, B. Sahu, P. Tarakeshwar, K.S. Kim, J. Phys. Chem. B 107 (2003) 9994-10005.

[3] H. Adachi, K. Taki, S. Nagamine, A. Yusa, M. Ohshima, J. Supercrit. Fluids 49 (2009) 265-270.

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